

# Modeling chemistry in protoplanetary disks

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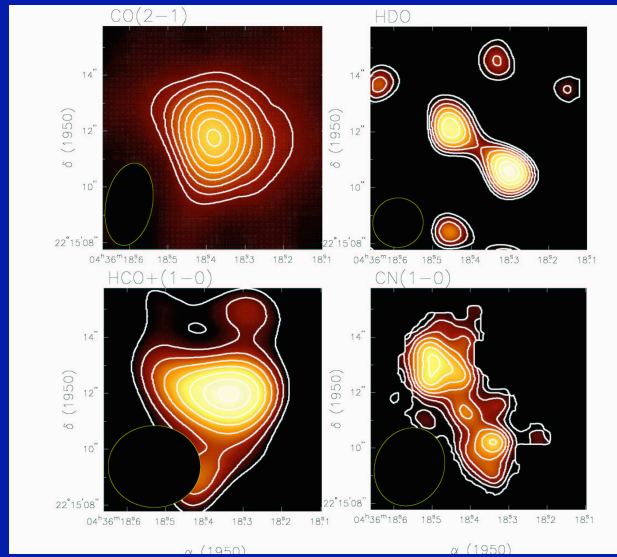
## Outline

- Two models
  - Deuterium chemistry  
Important to trace the thermal history of the disk and interstellar/disk links
  - Vertical mixing  
Can turbulent mixing in the vertical direction affect column densities in the disk

## Observations of disks

- Mostly sensitive to outer disk ( $> 50$  AU)
- Trace region with  $T = 20 - 40$  K and  $n = 10^6 - 10^8 \text{ cm}^{-3}$
- Fractional abundances are lower than in molecular clouds (factor of 10 - 100).
- Important processes
  - Ion-molecule e.g.  $\text{HCO}^+$
  - Photoprocesses e.g. CN/HCN ratio
  - High deuteration ratios
  - Low abundances of complex species e.g.  $\text{H}_2\text{CO}$ ,  $\text{CH}_3\text{OH}$

## LkCa15



T Tauri star  
 Distance = 140pc  
 Dust mass =  $10^{-4} - 0.2 M_{\odot}$   
 Age = 3 - 5 Myrs  
 Radius = 600 AU

Kessler 2003

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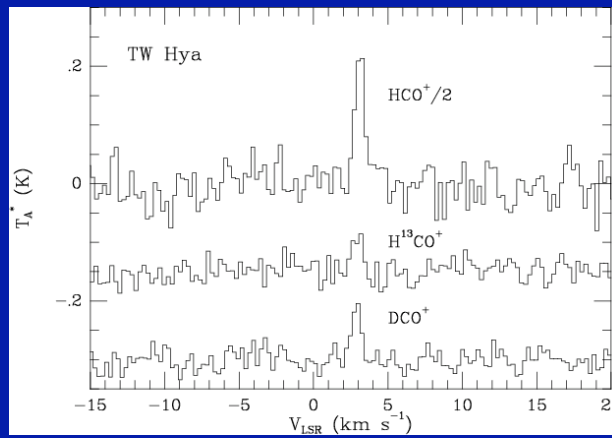
## OVRO observations of LkCa15

HCN - traces outer region with dip in column density in the center

CN - indicates photodissociation processes

HCO+ - indicates ion-molecule reactions - central peak

## TW Hya



T Tauri

Distance = 56pc

Age = 8 Myrs

Radius = 800 AU

Dust mass =  $2 \times 10^{-2} M_{\odot}$

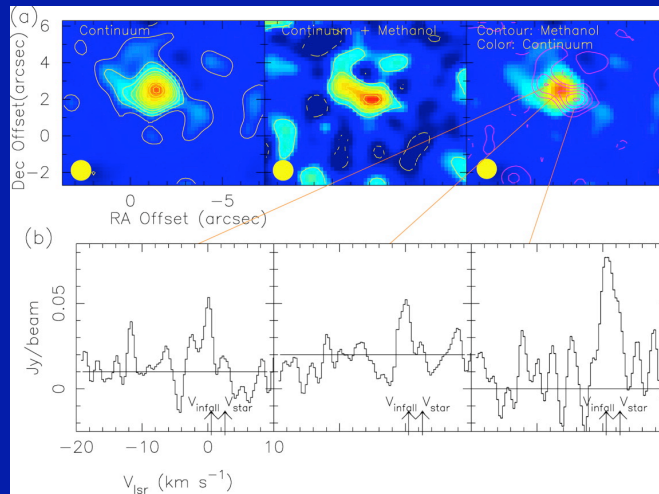
Van Dishoeck et al. (2003)

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## L1157



Velusamy, Langer & Goldsmith 2002

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Maps of dust continuum and methanol spectral line emission at 1mm from disk in L1157

Color image - continuum

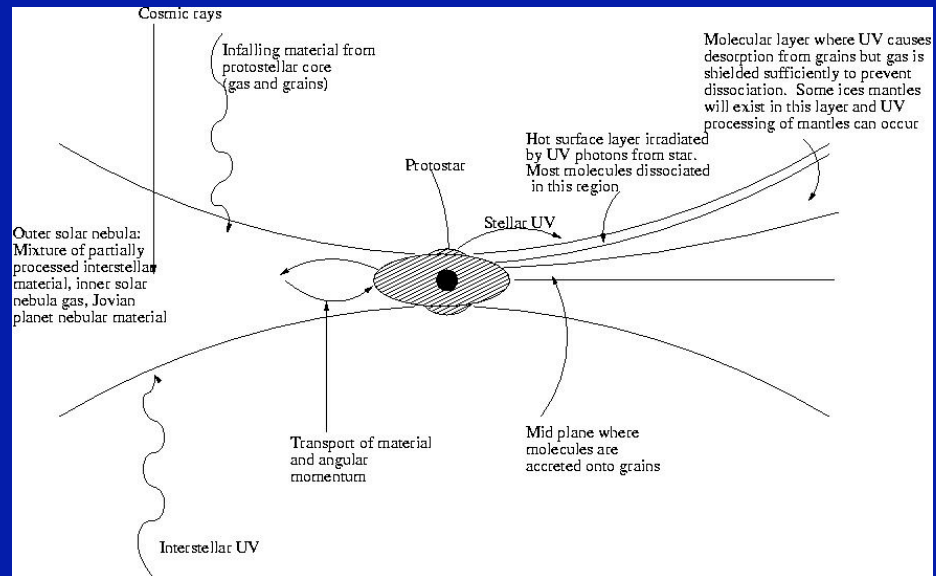
Line = CH<sub>3</sub>OH

Structure and kinematics of methanol emission indicate that the gas is tracing a warm layer in the infall-disk interface, consistent with an accretion shock,

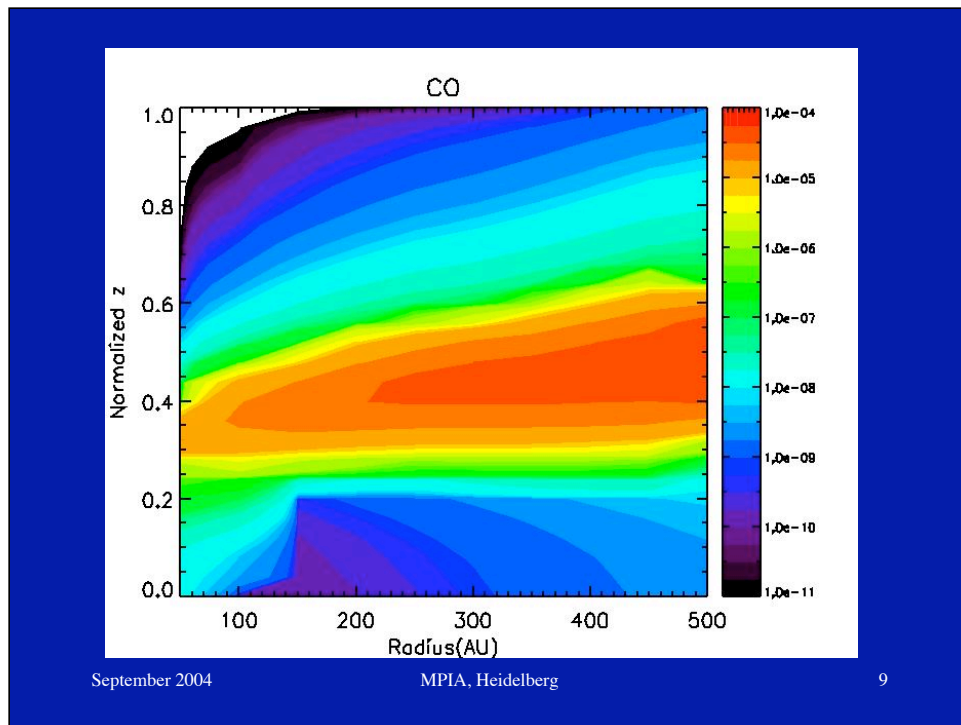
## Previous flared disk models

- Aikawa et al. (1999, 2001)
  - Kyoto minimum solar mass nebula model
  - Low temperature, isothermal model, requires low sticking coefficient ( $S_x = 0.03$ ) to maintain gas phase molecules
- Willacy & Langer (2000)
  - Chiang & Goldreich (1997) 2 layer model
  - Require efficient desorption process to keep molecules in the gas (photodesorption)
- Aikawa et al. (2002), van Zadelhoff et al. (2003)
  - D'Alessio et al. (1999) models with continuous vertical T gradient
  - Disk is warm enough that thermal desorption is efficient and can produce high abundances of molecules in the gas even for  $S_x = 1$

## Chemical structure of disks



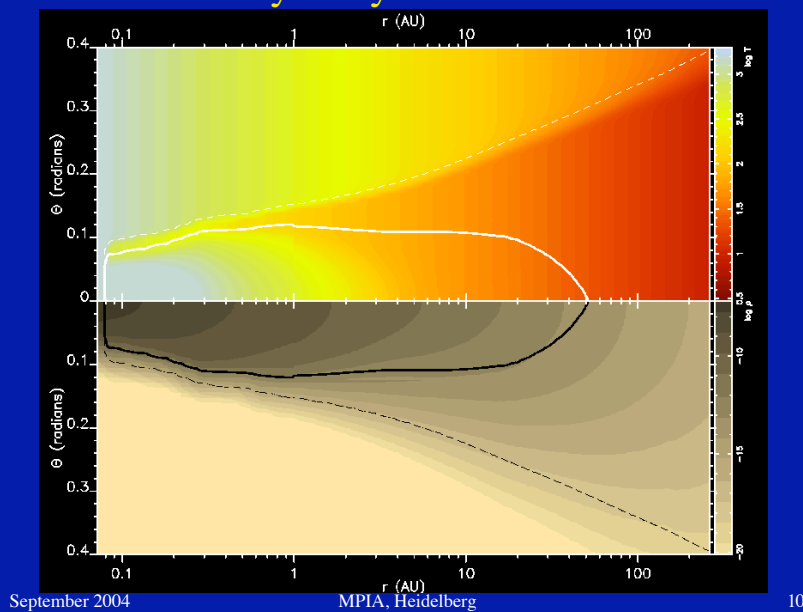




Fractional abundance of CO in disk

Shows 3 layer structure, with CO molecule existing in band, above the mid-plane and below the UV irradiated surface layer

## The hydrodynamical model



Bryden et al.

Describes a geometrically thin disk

Heated viscously and by radiation from central star

Temperature is calculated self-consistently

Assumes disk is in steady state I.e. disk properties are independent of time and mass accretion rate is constant

Structure of disk is determined by solving hydrodynamic equations in a semi-analytic fashion

Mass accretion rate =  $10^{-8}$  solar masses per year

Surface density =  $1000 \text{ g cm}^{-2}$  at 100AU

Surface density =  $\sigma_0 R^{-3/2}$   $R > 16 \text{ AU}$

Surface temperature proportional to  $R^{-3/7}$

Top half of plot is temperature - shows heated surface layer

Bottom half is density

Solid dark line is surface of disk based on vertical depth of its own infrared radiation

Dotted line is optical surface to radial visible stellar radiation

Disk is stable if

$H/r > M_{\text{disk}}/M_{\text{star}}$

## The chemical model

- Based on Millar et al. (1997) (UMIST ratefile)
- 268 species (gas and grain), 5312 reactions
- Include thermal desorption, cosmic ray heating and photodesorption
- Deuterium reactions (based on Willacy & Millar 1998, Rodgers & Millar 1996)
- Include H<sub>2</sub> and CO self-shielding using method of Lee et al. (1996).
- Photodissociation by both interstellar and stellar UV photons
- Input abundances from the output of a molecular cloud model at 1Myrs

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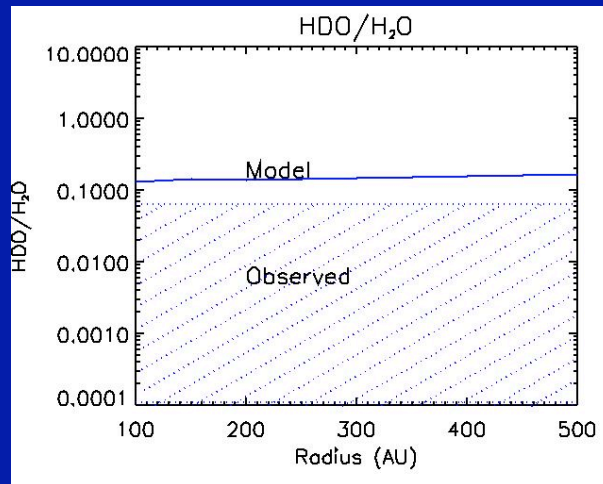
### Desorption processes -

Thermal - material desorbs as temperature reaches its sublimation temperature so strongly bound molecules desorb at higher temperatures

Cosmic ray heating - not efficient enough for most molecules to make a difference

Photodesorption - included to represent some kind of efficient desorption, the cooler temperatures in our model which arise from the use of a more realistic grain size distribution (instead of an interstellar distribution) means that thermal desorption is not sufficient on its own to maintain observable gas phase column densities of many species. Photodesorption is potentially very efficient if the lab work of Westley et al. (1995) is correct but we use it here mainly as a representative desorption process rather than claiming that the extra desorption process has to be photodesorption (desorption processes are a matter of contention!).

## Deuterium model results



- HDO/H<sub>2</sub>O  
10<sup>-4</sup> - 0.064  
(Kessler 2003)

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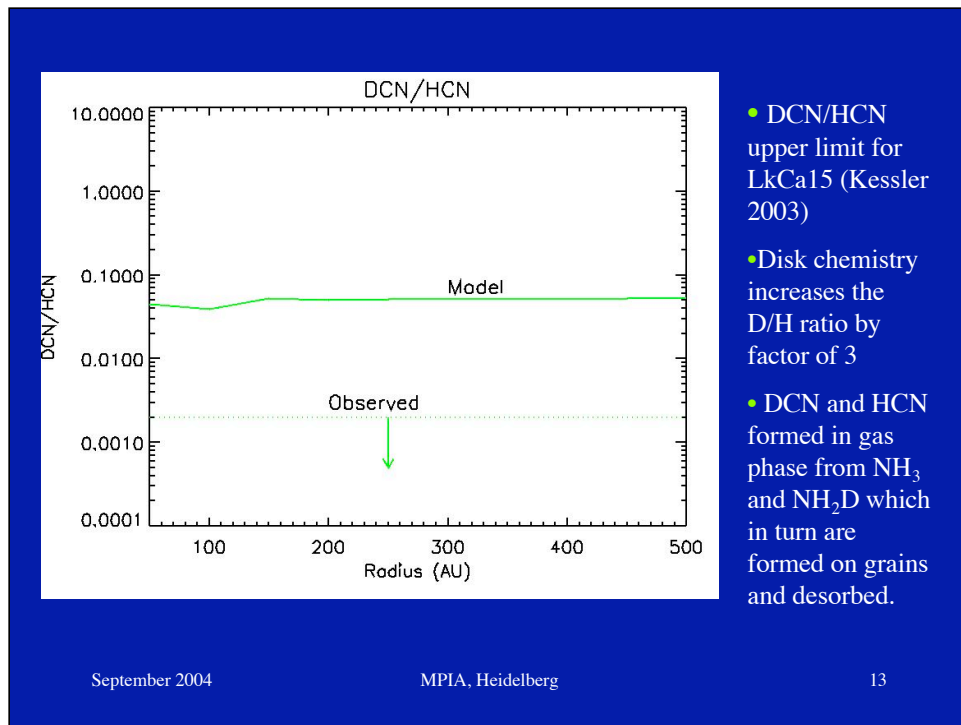
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General comments - compared to observations of DM Tau, LkCa15, TW Hya  
Differences in physical parameters of model compared to sources could affect the comparison e.g. disk mass - higher disk masses could result in higher column densities. But in general we find fairly good agreement with many molecules agreeing with observations to a factor of a few. Given uncertainties in the models this is good to see!

Upper limit observed = 0.0001 - 0.064

Model ~ 0.2

HDO/H<sub>2</sub>O is determined by grain chemistry and desorption.



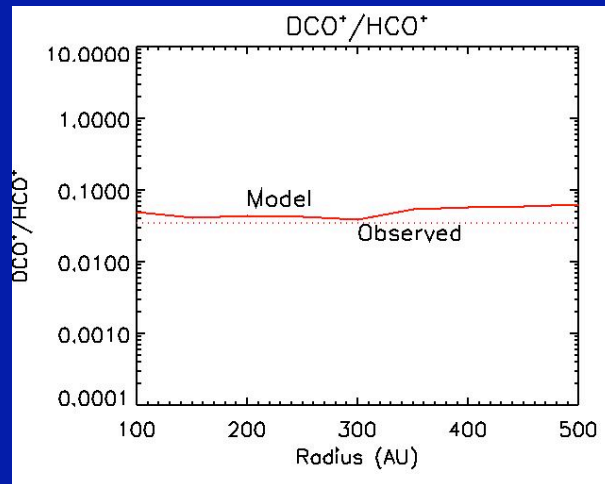
Observed value in disk is less than the observed value in TMC-1 (value = 0.023).

Model input value is 0.014 close to TMC-1 value.

Model increases ratio, but observations suggest that the ratio is decreased in disks

DCN/HCN ratio therefore reflects the grain D/H abundance of  $\text{NH}_2\text{D}/\text{NH}_3$

May indicate that the grain chemistry formation of  $\text{NH}_3$  etc is too efficient



- Observed value from van Dishoeck (2004) (TW Hya)
- Good agreement with observations
- Formation in gas from  $\text{HCO}^+ + \text{D} \rightarrow \text{DCO}^+ + \text{H}$

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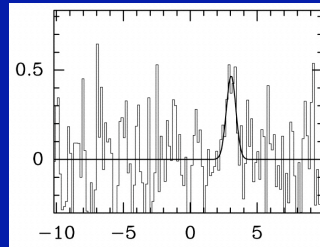
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Ratio determined by abundance of atomic D.

## $\text{H}_2\text{D}^+$

TW Hya

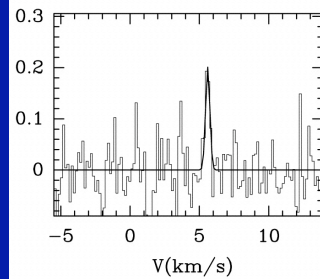


TW Hya:  $x(\text{H}_2\text{D}^+) = 3.4 \times 10^{-10}$

DM Tau:  $x(\text{H}_2\text{D}^+) = 4.2 \times 10^{-10}$

Model:  $x(\text{H}_2\text{D}^+) = 8.7 \times 10^{-11}$

DM Tau



Ceccarelli et al. 2004

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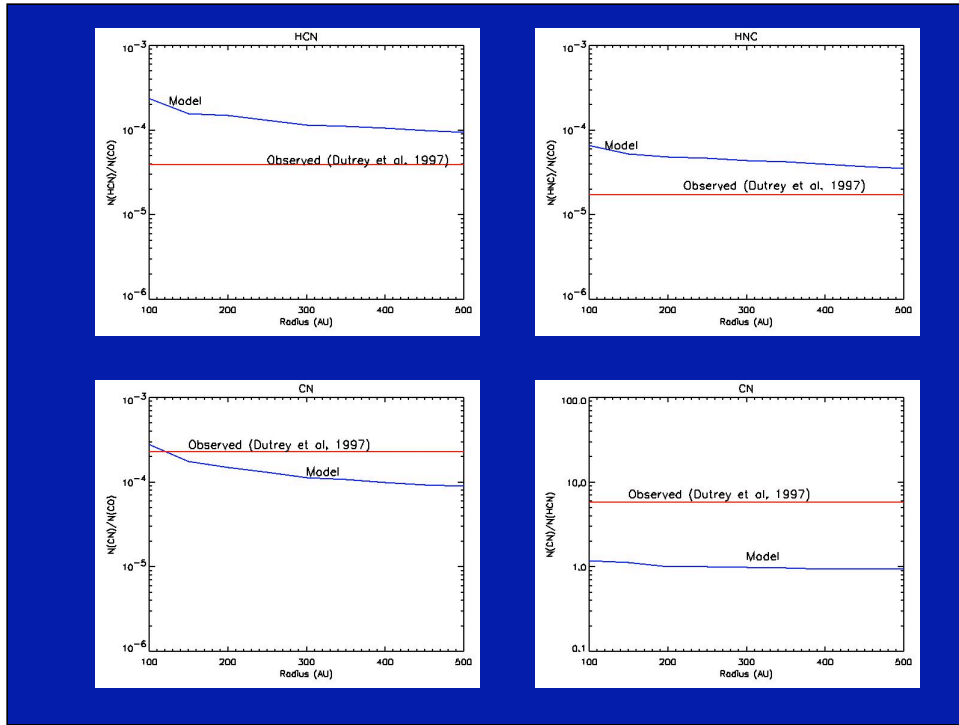
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Observations thought to trace the midplane where the gas is cold and depleted of CO.

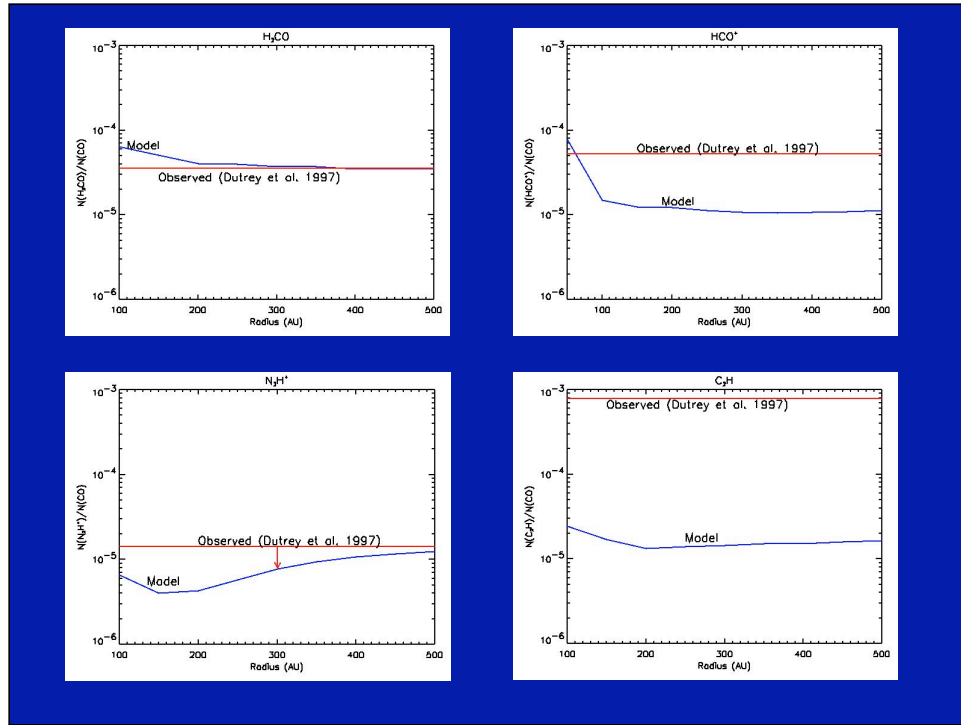
Model value is for 300AU at midplane.

Agreement within a factor of 5 - pretty good!



Good agreement (within a factor of a few) for many molecules in outer disk  
 CN/HCN a bit on the low side - could be due to low stellar UV field, or to the lack of proper treatment of the radiative transfer





Good agreement in outer disk except for C<sub>2</sub>H - again photodissociation product, could be problems with radiative transfer or the low UV field from the star.

## Results

- Successes -
  - Good agreement with many observed abundances including the D/H ratios for  $\text{HCO}^+$  and  $\text{H}_2\text{O}$
- Problems
  - Can't account for the observed abundances of molecules such as  $\text{C}_2\text{H}$  and  $\text{CN}$  which are formed by photodissociation. Need a better description of the radiative transfer in the disk

## Diffusion model

- Disks are turbulent
  - Efficient angular momentum transport required
  - Physical conditions in disks    turbulent transport only process that can achieve this transport
- Shakura & Sunyaev (1973)
  - Introduced an ‘anomalous’ viscosity component proportional to gas pressure
- Lyndon & Pringle (1974)
  - $\nu_t = \alpha c_s h$

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Efficient angular momentum transport required - molecular viscosity is not efficient enough. Therefore turbulence was invoked. Source of this turbulence is still uncertain, could be MHD, hydrodynamical or gravitational.

Pringle - from this can construct models of vertically averaged alpha disks.

$\nu_t$  = turbulent viscosity

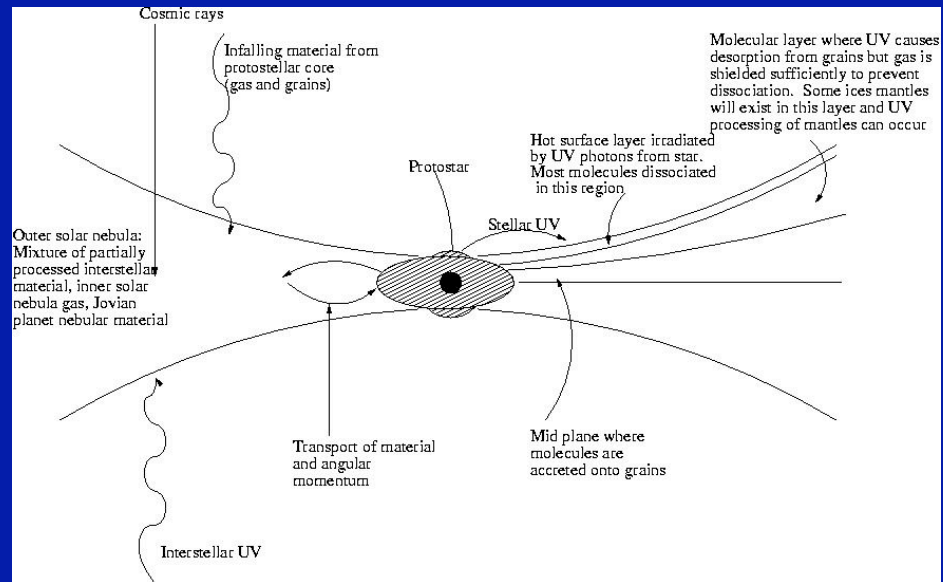
$\alpha$  = dimensionless parameter  $< 1$

$c_s$  = sound speed

$H$  = disk height

From this models of vertically average alpha disks can be constructed.

## Why diffusion might affect chemistry



- Code originally used for atmospheric chemical models (Allen, Yung & Waters 1981, Shia et al. 1990), and for modeling molecular clouds (Xie, Allen & Langer 1995, Willacy, Langer & Allen 2002)
- Uses mixing length theory
- Assumes diffusion timescale for given tracer depends on its composition gradient

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Mixing length theory - characterises turbulence as eddies that maintain different properties to the average fluid in which they exist for the time taken to travel a distance  $l$  - the mixing length

Eddies remain separate to the ambient fluid for long enough to travel a distance  $l$  - the mixing length

## 1-D diffusion

Fluctuations in fractional abundance  $x_i$  due to turbulence can be parameterized as

$$x_i = l \frac{dx_i}{dz}$$

Net transport flux:

$$F_i = n(H_2) \langle v_t x_i \rangle = -Kn(H_2) \frac{dx_i}{dz}$$

$$F_i = -Kn_i \frac{1}{n_i} \frac{dn_i}{dz} = -\frac{1}{n(H_2)} \frac{dn(H_2)}{dz}$$

$K$  = diffusion coefficient =  $\langle v_t l \rangle$

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$dx_i/dz$  = abundance gradient in the  $z$  direction,  $l$  = mixing length

$V_t$  = turbulent viscosity

- Chemical continuity equations

$$\frac{dn_i}{dt} + \frac{\Phi_i}{z} = P_i - L_i$$

- Turbulent code solves coupled continuity equations using finite differencing method
- Code allows different temperatures, densities, diffusion and advection coefficients in each zone

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P, L = production and loss terms of species I

Phi\_I = net transport flux

## Estimating K

- Estimate diffusion coefficient K by

$$K = \frac{R^2}{t}$$

$$t = c_s h$$

For 100 AU

$\alpha = 0.001$ ,

$h = 0.09 \times 100 \text{ AU} = 1.35 \times 10^{14} \text{ cm}$

$c_s = 4.4 \times 10^3 \text{ cm s}^{-1}$

$K = 6 \times 10^{15} \text{ cm}^2 \text{ s}^{-1}$

- We use K between  $10^{18}$  and  $10^{20} \text{ cm}^2 \text{ s}^{-1}$
- We assume K constant with R and z

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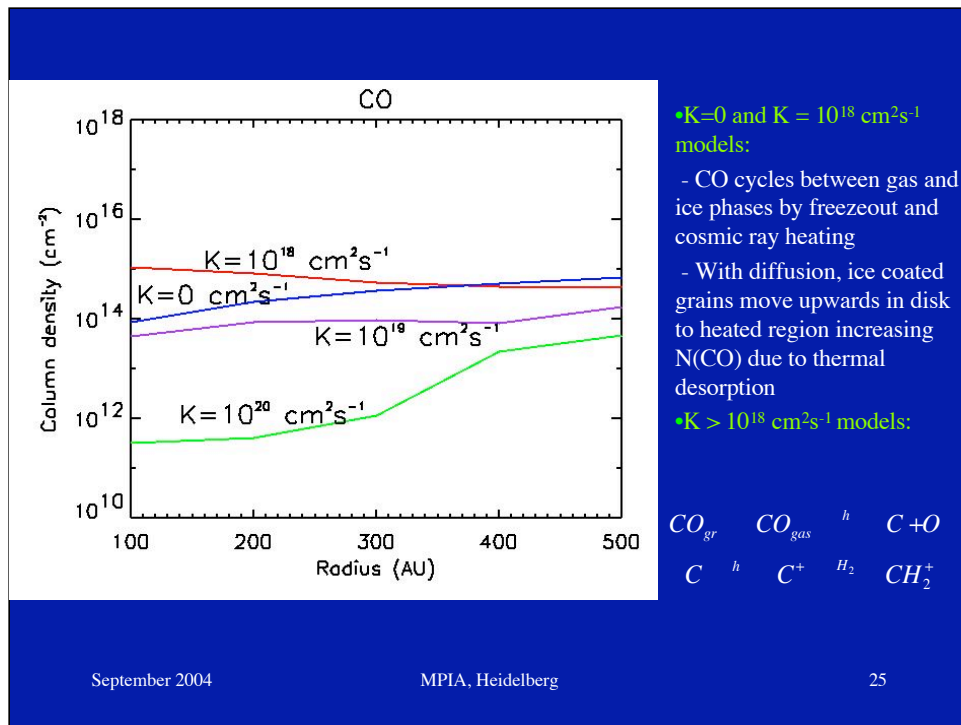
$c_s$  = sound speed

H = scale height

$\alpha$  = alpha parameter used to describe viscosity in disks

Computational problems with  $K < 10^{18}$





Low levels of diffusion increase column density at small radii, but higher values actually decrease it.

At 200 AU and 1 Myr:

CO ice forms in midplane. In  $K=0$  model, it is desorbed by CR heating and cycles between gas and grain phases.

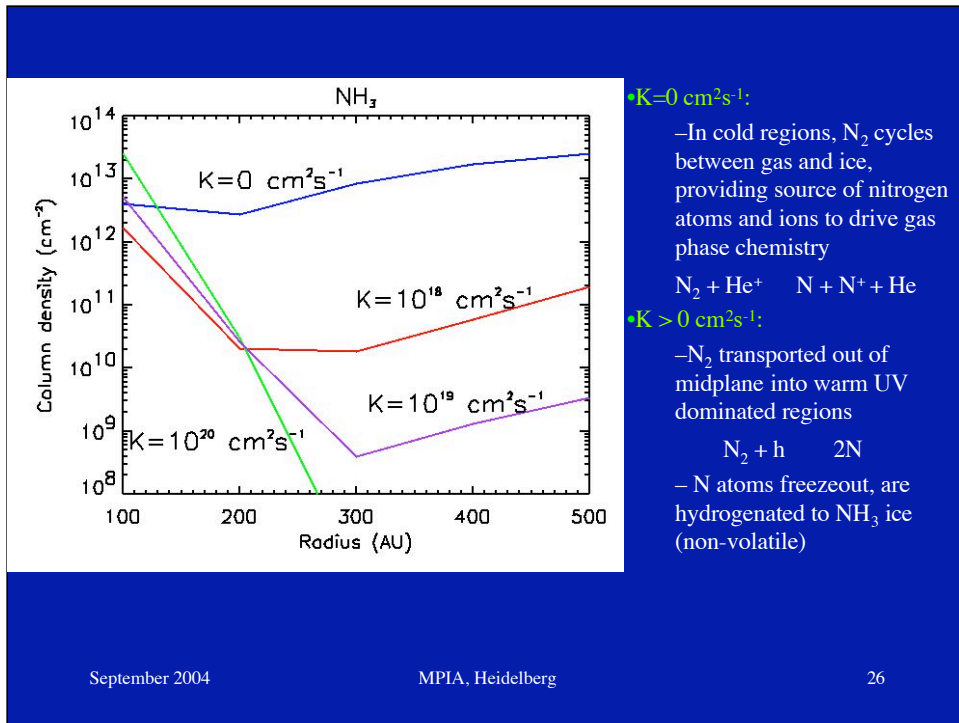
In  $K=10^{18}$  model, ice coated grains transported out, CO desorbed thermally, cycles between gas and grain - get ice and gas at higher  $z$  than in  $K=0$  model

For higher values of  $K$

Ice coated grains transported outwards, CO desorbed, but in outer layers it is also photodissociated into C and O. C is photoionized to  $\text{C}^+$ , which reacts with  $\text{H}_2$  to start chain that forms  $\text{CH}_4$ . Therefore CO is gradually changed into  $\text{CH}_4$ . For  $K=10^{20}$  model,  $\text{CH}_4 \sim 5\times$  higher in midplane than  $x(\text{CH}_4)$  in  $K=10^{18}$  model and extends to higher  $z$ .

Low  $K$  - thermal desorption in upper layers balances accretion in lower layers so CO abundance in gas remains high. Little processing into other molecules

High  $K$  - CO dissociated in upper layers, converted into carbon atoms and then into hydrocarbons such as methane



Diffusion reduces column densities at all values of K

All nitrogen bearing molecules show reduced column densities with the addition of diffusion for  $R > 150 \text{ AU}$ .

In  $K=0$  model, NH<sub>3</sub> is present in the midplane region  $z < 1.4 \times 10^{15} \text{ cm}$

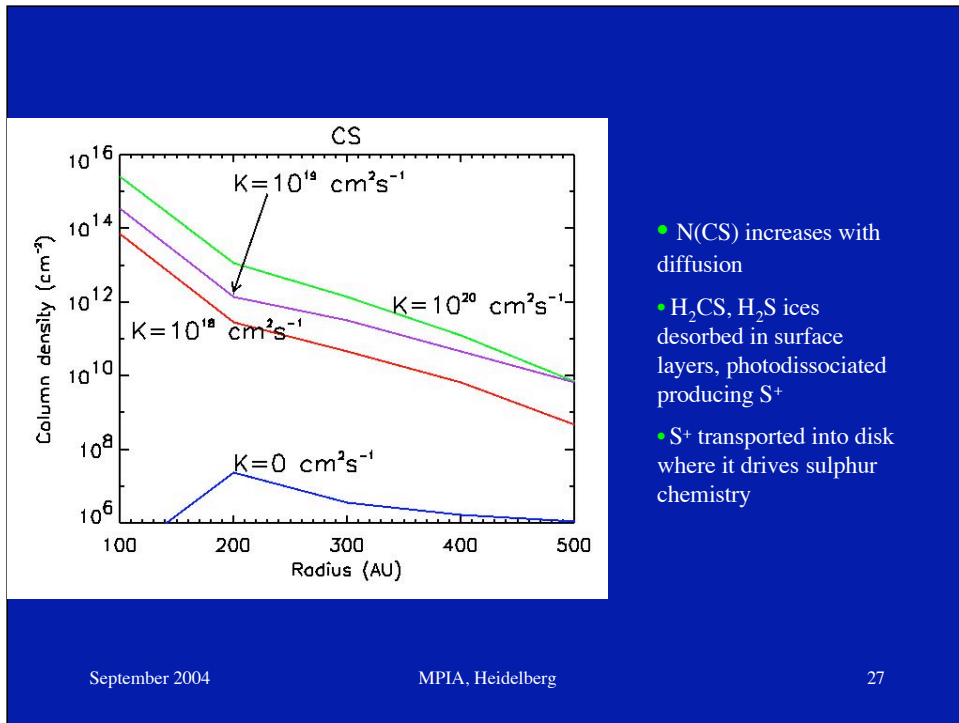
It forms from desorption of N<sub>2</sub> from grains, which reacts with He<sup>+</sup>, producing N and N<sup>+</sup> which drive N chem

In outer region, N<sub>2</sub> desorbs, is photodissociated into 2 N atoms, which then freezeout and form NH<sub>3</sub> on grains, without much being left in gas therefore little gas phase chemistry. N ends up in NH<sub>3</sub> ice which is non-volatile and remains on grains (photodesorption not included here).

So in  $K>0$  models, N<sub>2</sub> processed into NH<sub>3</sub> ice in outer regions, transported into lower levels where it can't be desorbed and there is no reservoir of N atoms for gas phase chemistry. Therefore N ends up trapped on grains in NH<sub>3</sub> ice and N(NH<sub>3</sub>) gas phase column density falls.

At small radii:

NH<sub>3</sub> forms on grains in cold mid-plane. Diffusion moves the grains into warm region where NH<sub>3</sub> is desorbed thermally. Hence there is a warm layer containing high abundances of NH<sub>3</sub> in the gas. Increasing K increases the movement of NH<sub>3</sub> ice from the midplane upwards



Column densities of CS are increased with diffusion.

Dissociation of sulphur molecules in upper layers (e.g.  $\text{H}_2\text{CS}$ ,  $\text{H}_2\text{S}$  which are partially desorbed - thermally,  $\text{BE} = 2000\text{K}$  for  $\text{H}_2\text{S}$  and  $2250\text{K}$  for  $\text{H}_2\text{CS}$  so there is some desorption in warm surface layers). This provides source of sulphur atoms.

Diffusion allows  $\text{S}^+$  to be transported into disk where it can drive a sulphur chemistry, resulting in an increase in the CS abundance in the molecular layer.

Increasing  $K$  increases the thickness of the layer that contains  $\text{S}^+$  and therefore sulphur molecules, so sulphur molecules are present deeper into the disk.

## Conclusions

### Mixing model:

- Mixing can greatly affect the abundances
- Effects of mixing are molecule dependent.

### Deuterium model:

- Good agreement with observations for many molecules in outer disk
- Good agreement for D/H ratios of  $\text{H}_2\text{O}$  and  $\text{HCO}^+$
- Good agreement with midplane abundance of  $\text{H}_2\text{D}^+$
- Some problems with photodissociation products - need better description of radiative transfer

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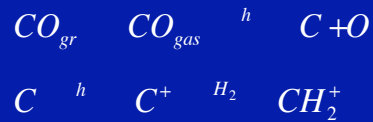
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Some molecules show increase in column density, others are decreased, mixing greatly complicates the chemical processing.

## CO chemistry

- $K=0$  and  $K = 10^{18} \text{ cm}^2\text{s}^{-1}$  models:
  - CO cycles between gas and ice phases by freezeout and cosmic ray heating
  - With diffusion, ice coated grains move upwards in disk to heated region. Thermal diffusion and freezeout balance to form a CO peak at higher  $z$  than for  $K=0$  model
- $K > 10^{18} \text{ cm}^2\text{s}^{-1}$  models:



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Low  $K$  - thermal desorption in upper layers balances accretion in lower layers so CO abundance in gas remains high. Little processing into other molecules

High  $K$  - CO dissociated in upper layers, converted into carbon atoms and then into hydrocarbons such as methane

## Observations

- Use observations of LkCa15 (Kessler 2003, Qi 2001), DM Tau (Dutrey et al. 1997) and TW Hya (van Dishoeck et al. 2004).
- Observations from outer disk
- LkCa15: Dust mass =  $0.2 M_{\odot}$ , age = 3-5 Myrs,  
radius = 600AU
- DM Tau: Mass = 0.03

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Observations from outer disk -  $R > \sim 100$  AU

Therefore some differences between model and observations would be expected because of different physical parameters - future models will consider specific sources.

Aikawa & Herbst 2001 - found that higher disk masses lead to higher D/H ratios